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# SiO<sub>2</sub> passivation layer grown by liquid phase deposition for silicon solar cell application

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**Abstract** Surface passivation is one of the primary requirements for high efficient silicon solar cells. Though the current existed passivation techniques are effective, expensive equipments are required. In this paper, a comprehensive understanding of the SiO<sub>2</sub> passivation layer grown by liquid phase deposition (LPD) was presented, which was cost-effective and very simple. It was found that the post-annealing process could significantly enhance the passivation effect of the LPD SiO<sub>2</sub> film. Besides, it was revealed that both chemical passivation and field-effect passivation mechanisms played important roles in outstanding passivation effect of the LPD SiO<sub>2</sub> film through analyzing the minority carrier lifetime and the surface recombination velocity of n-type and p-type silicon wafers. Although the deposition parameters had little influence on the passivation effect, they affected the deposition rate. Therefore, appropriate deposition parameters should be carefully chosen based on the compromise of the deposition rate and fabrication cost. By utilizing the LPD SiO<sub>2</sub> film as surface passivation layer, a 19.5%-efficient silicon solar cell on a large-scale wafer (156 mm × 156 mm) was fabricated.

**Keywords** Si solar cell, passivation, SiO<sub>2</sub>, liquid phase deposition, carrier lifetime

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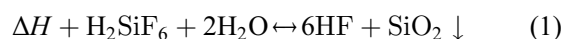
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## 1 Introduction

It is well known that surface passivation plays a key role in high-efficiency silicon solar cells. SiO<sub>2</sub> film is proved to be one of the most effective dielectric passivation layers and thus gains lots of concerns in photovoltaic industry [1,2]. In general, the SiO<sub>2</sub> passivation layer is thermally grown in a tube furnace or deposited by vacuum-based systems such as the expanding thermal plasma technique [3,4]. Although the SiO<sub>2</sub> film deposited by these approaches can effectively passivate the silicon surface, especially after annealing in forming gas [3,4], none of these methods is ideal for low-cost manufacturing. Considering the cost-effectiveness, attempts are made to find another approach to produce the high-quality SiO<sub>2</sub> film with low cost. Fortunately, the liquid phase deposition (LPD) for growing SiO<sub>2</sub> film may provide a good solution with many advantages such as low cost, simple equipment, high deposition rate and low processing temperature.

The mechanism of the LPD was originally proposed by Nagayama et al. [5]. In briefly, the deposition processing of SiO<sub>2</sub> is based on the hydrolysis reaction of silica-supersaturated hydrofluosilicic acid written as



where  $\Delta H$  means that the positive reaction is an endothermic reaction. When the reactants (hydrofluosilicic acid and water) are excessive, the equilibrium reaction will shift from left to right, resulting in the precipitation of SiO<sub>2</sub>. In recent years, there are lots of reports on LPD SiO<sub>2</sub> film [6–14]. Most of the reports focus on investigating the effect of growth conditions involving the growth temperature, the solution concentration and the growth time on the characters of the LPD SiO<sub>2</sub> film. Up to the present, the LPD SiO<sub>2</sub> film is mainly applied to microelectronics as dielectric insulating layer [10–12] but is rarely applied to solar cells as a passivation layer. Yuan et al. [13] have first tried to apply LPD SiO<sub>2</sub> film to silicon solar cells as the passivation layer and fabricated a 16.4%-efficient black

silicon solar cell, presenting the promising application of LPD SiO<sub>2</sub> film in silicon solar cells. He et al. [14] have also applied LPD SiO<sub>2</sub> passivation layer to multicrystalline silicon solar cells, but the passivation effect and the resulted conversion efficiency (5.61%) are not satisfying. Therefore, it is of significant importance to comprehensively understand the LPD SiO<sub>2</sub> passivation layer and propose the approach to realize outstanding passivation effect.

In this paper, the influence of the deposition conditions together with post-annealing treatment on the properties and passivation effect of the LPD SiO<sub>2</sub> film was systematically investigated. It was found that an annealing temperature higher than 700°C was required to obtain a high-quality passivation effect of the LPD SiO<sub>2</sub> film. Moreover, the film had an excellent passivation effect on both p-type and n-type silicon wafers but a better passivation effect on n-type, suggesting the simultaneous presence of chemical passivation and field-effect passivation. Though the deposition conditions did not have much influence on the passivation effect of the LPD SiO<sub>2</sub> film, they affected the deposition rate. A medium concentration of H<sub>2</sub>SiF<sub>6</sub> (1.5 M) and a medium temperature (50°C) were suggested to be the optimal deposition conditions based on the tradeoff of material cost, thermal cost and deposition rate. This optimized LPD SiO<sub>2</sub> film was successfully applied to silicon solar cells as the front surface passivation layer, achieving a conversion efficiency of 19.5% on a large scale (156 mm × 156 mm).

## 2 Experimental details

### 2.1 Deposition of SiO<sub>2</sub> film

The whole deposition system mainly consists of a Teflon beaker containing a silica-supersaturated hydrofluosilicic acid (H<sub>2</sub>SiF<sub>6</sub>) solution, a holder for holding silicon wafers and a magnetic stirring apparatus (90-18, Shanghai Sile Instrument Co., Ltd.). The solution temperature was controlled by using the water bath method. To prepare the silica-supersaturated H<sub>2</sub>SiF<sub>6</sub> solution, silicic acid powders (99.9%, Sigma-Aldrich) were added to a raw H<sub>2</sub>SiF<sub>6</sub> solution (33.5%–35%, Sigma-Aldrich) and stirred at 400 r/min for 20 min at room temperature to obtain the silica-saturated H<sub>2</sub>SiF<sub>6</sub> solution. The concentration of H<sub>2</sub>SiF<sub>6</sub> in the silica-saturated solution was 3.09 M. Then, an appropriate amount of deionized (DI) water was added to the silica-saturated H<sub>2</sub>SiF<sub>6</sub> solution to form the silica-supersaturated solution which was directly used for the SiO<sub>2</sub> deposition. The concentration of H<sub>2</sub>SiF<sub>6</sub> in the supersaturated solution was controlled by the amount of the DI water added. The substrates utilized for SiO<sub>2</sub> deposition were Czochralski (Cz) silicon with a resistivity of 1–3 Ω·cm for p-type and 3–4 Ω·cm for n-type and a thickness of 200 μm. Before depositing SiO<sub>2</sub> film, the

substrates were pretreated by being dipped into diluted HF solution and diluted HCl solution to remove the natural silicon dioxide and metallic ions, and then immersed into DI water to gain OH-terminated surface, which substantially facilitated the growth of the SiO<sub>2</sub> film [6]. After these pretreatments, the silicon wafers were immersed in the supersaturated solution with the controlled concentration of H<sub>2</sub>SiF<sub>6</sub> and temperature to deposit SiO<sub>2</sub> films.

### 2.2 Fabrication of silicon solar cells

For the fabrication of silicon solar cells, the crystalline silicon was Cz p-type with a resistivity of 1–3 Ω·cm, a thickness of 200 μm and an area of about 238.95 cm<sup>2</sup> (quasi-square: 156 mm × 156 mm). After surface texturization and p-doping steps, the phosphorosilicate glass (formed in the doping process) on the rear side was removed in the HF solution first. Then, the silicon wafers were immersed in tetramethylammonium hydroxide solution to produce polished rear surface. Next, the phosphorosilicate glass on the front side was removed and the SiO<sub>2</sub> films were deposited on both sides of the wafers by the LPD. The wafers were annealed in a muffle furnace at the annealing temperature of 800°C for 5 min to obtain a high-quality passivation effect and subsequently the SiO<sub>2</sub> film on the back side was removed by using the HF. After that, SiN<sub>x</sub> coatings were deposited on the SiO<sub>2</sub> films as the antireflection layers by plasma enhanced chemical vapor deposition. Finally, the electrodes were fabricated on both the front and back sides by screen printing technique and co-fired in firing furnace.

### 2.3 Characterizations

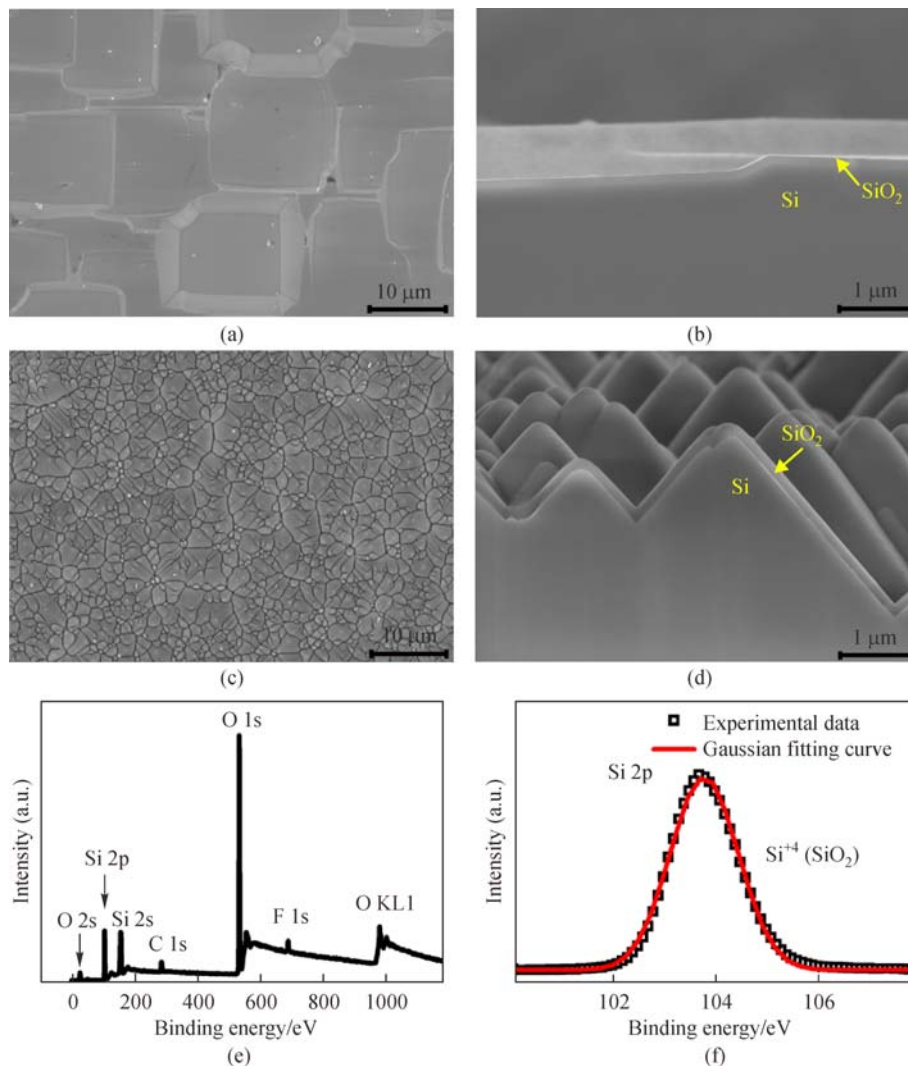
The effective minority carrier lifetime ( $\tau$ ) of the Si wafers symmetrically passivated by the LPD SiO<sub>2</sub> film was measured by using the Semilab WT-1200A lifetime tester, which directly reflected the passivation quality. The thermolyne muffle furnace was used to study the impact of post-annealing on the surface passivation effect of the LPD SiO<sub>2</sub> films. The surface morphologies of the LPD SiO<sub>2</sub> films on different substrates were observed by using the scanning electron microscope (SEM). The element compositions and chemical configuration of the LPD SiO<sub>2</sub> films were characterized by using the X-ray photoelectron spectroscopy (XPS), and using monochromatic Al K<sub>α</sub> radiation as the X-ray source. The infrared absorption spectra were measured by using the Fourier transform infrared (FTIR) spectrometer in the range of 350 to 8000 cm<sup>-1</sup>. The refractive index and thickness of the LPD SiO<sub>2</sub> films were determined by using the spectroscopic ellipsometry. The current ( $I$ )–voltage ( $V$ ) tester was used to measure the electrical performances ( $I$ – $V$  and power ( $P$ )– $V$  curves) of the LPD SiO<sub>2</sub> passivated solar cell under AM1.5 spectrum at the temperature of 25°C.

### 3 Results and discussion

#### 3.1 Quality of LPD SiO<sub>2</sub>

SiO<sub>2</sub> films were successfully fabricated by the LPD under different growth conditions. For the morphological observation and the identification of the chemical composition, the films grown at 50°C for two hours with 1.0 M H<sub>2</sub>SiF<sub>6</sub> without loss of generality were chosen. Figures 1 (a) and (b) respectively show the top-view and cross-sectional SEM image of the LPD SiO<sub>2</sub> film deposited on the polished silicon substrate, both demonstrating that the SiO<sub>2</sub> film is uniformly and compactly deposited on the

silicon surface. The LPD SiO<sub>2</sub> film was also grown on the pyramid structure textured silicon surface which was the real circumstance for commercial silicon solar cells, shown in Fig. 1(c) and (d). The SiO<sub>2</sub> film was conformally deposited on the whole silicon pyramid textures in spite of the size of the pyramid, which was the primary requirement for effective surface passivation. To confirm the element composition and chemical configuration of the film, the samples above were characterized by using the XPS and FTIR spectrometer. Figure 1(e) depicts the XPS full-scan spectrum of the as-prepared LPD SiO<sub>2</sub> film. The peak values of elementary binding energy in the spectrum are all modified by standard C 1 s peak located at 284.6 eV.



**Fig. 1** Quality of LPD SiO<sub>2</sub> film

(a) Top-view of the LPD SiO<sub>2</sub> film grown on the polished silicon substrate; (b) cross-sectional SEM images of the LPD SiO<sub>2</sub> film grown on the polished silicon substrate; (c) top-view of LPD SiO<sub>2</sub> film grown on the pyramid structure textured silicon substrate; (d) cross-sectional SEM images of LPD SiO<sub>2</sub> film grown on the pyramid structure textured silicon substrate; (e) XPS full-scan spectrum of the as-deposited LPD SiO<sub>2</sub> film; (f) XPS Si 2p spectrum of the as-deposited LPD SiO<sub>2</sub> film (The black open squares represent the experimental data and the red solid curve is the Gaussian fitting curve based on the raw data.)

The peaks of 25.3 eV, 103.68 eV, 153.96 eV, 284.6 eV, 532.98 eV, 688.98 eV and 979.5 eV are respectively corresponding to O 2s, Si 2p, Si 2s, C 1s, O 1s, F 1s and O KL1 [15–17]. The carbon detected by XPS may have been induced by the external circumstance (e.g., engine oil of XPS instrument and atmosphere). The fluoride (F) comes from the H<sub>2</sub>SiF<sub>6</sub> reactant, which exists in the form of Si-F

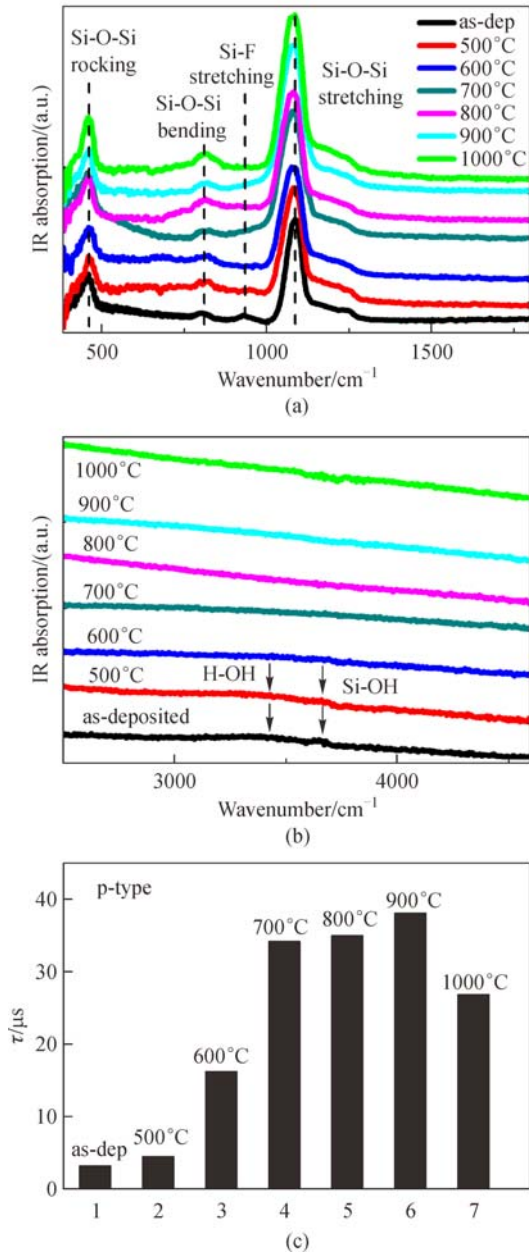
bond in the film as displayed in Fig. 2(a). Other impurities are below the detection limit. Figure 1(f) shows the XPS fine-scan spectrum for Si 2p. The XPS experimental data shows a symmetric peak, which fits well with a single Gaussian peak located at 103.77 eV, revealing the chemical state of the LPD SiO<sub>2</sub> is silicon dioxide without any suboxide [18].

Figure 2((a) and (b)) further displays the FTIR absorption spectra of the as-prepared LPD SiO<sub>2</sub> film (black curves). The characteristic peaks at 457 cm<sup>-1</sup>, 800 cm<sup>-1</sup> and 1085 cm<sup>-1</sup> are corresponding to the Si-O-Si rocking vibration mode, the Si-O-Si bending vibration mode, and the Si-O-Si stretching vibration mode, respectively [10,11,19]. The absorption peak at 931 cm<sup>-1</sup> may be attributed to the Si-F stretching vibration mode [10,11]. Additionally, as exhibited in Fig. 2(b), the infrared absorption signal in the range of 3000 cm<sup>-1</sup> to 4000 cm<sup>-1</sup> is related to the OH-group vibration [10,19]. In detail, the weak peak at 3670 cm<sup>-1</sup> corresponds to the Si-OH stretching vibration mode while the absorption band at 3400 cm<sup>-1</sup> is attributed to the H-OH stretching vibration. No other prominent vibration modes are found in the spectrum from 350 cm<sup>-1</sup> to 8000 cm<sup>-1</sup>, well according with the result of XPS in Fig. 1.

### 3.2 Influence of annealing process

To study the effect of the post-annealing on the properties of the LPD SiO<sub>2</sub> films, the samples, which are the double-sided polished p-type Si wafers with the LPD SiO<sub>2</sub> films on both sides, were subjected to anneal at different temperatures in atmosphere for 5 min by muffle furnace. The annealed samples were also characterized by FTIR spectrometer, as shown in Fig. 2(a) and (b). Obviously, the signals of the Si-F, Si-OH and H-OH stretching vibration modes almost disappear after annealing at the temperatures higher than 700°C, contrasting to the gradually increased intensity of the Si-O-Si stretching vibration mode with the annealing temperature. The reason for this is that at a rather high annealing temperature, the Si-F bond and Si-OH bond break and transform to the Si-O-Si bond and HF, and the latter escapes away from the LPD SiO<sub>2</sub> films by evaporation. As a result, the element composition of the post-annealed LPD SiO<sub>2</sub> film is highly pure. In addition, compared with the spectrum of the as-deposited sample, the peak of the Si-O-Si stretching vibration mode slightly shifts to the lower wave number (red shift) after annealing. It is known that the Si-O-Si stretching vibration mode is closely related to the composition or the density of the SiO<sub>2</sub> film [20]. Therefore, it can be inferred that the increased density of the post-annealed SiO<sub>2</sub> film (which is verified by the increased refractive index, as shown in Fig. 3 and Fig. 4) should be responsible for the red shift of the Si-O-Si stretching vibration frequency in the study.

The passivation effect of the LPD SiO<sub>2</sub> films before and



**Fig. 2** Influence of annealing temperature on chemical bonds and passivation effect of LPD SiO<sub>2</sub> film

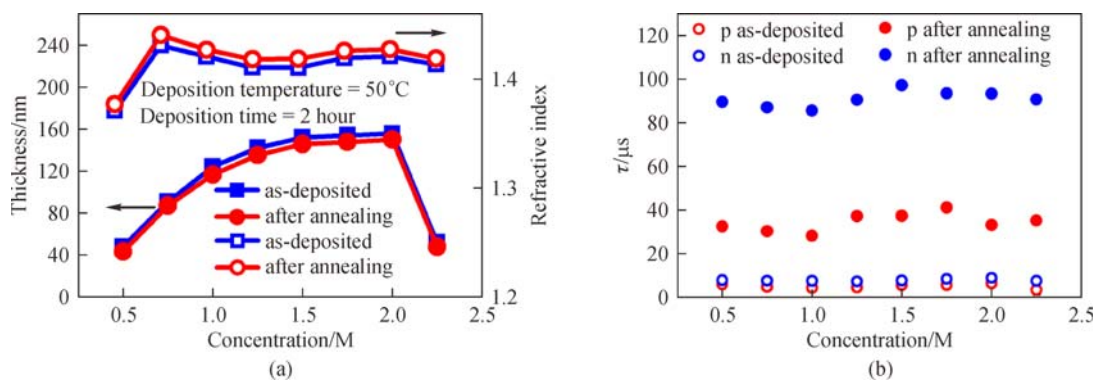
(a) FTIR spectra of the as-prepared LPD SiO<sub>2</sub> film and the LPD SiO<sub>2</sub> film annealed at different temperatures in the small wavenumber region; (b) FTIR spectra of the as-prepared LPD SiO<sub>2</sub> film and the LPD SiO<sub>2</sub> film annealed at different temperatures in the large wavenumber region; (c) histogram of  $\tau$  of the as-deposited and post-annealed samples (as-dep represents as-deposited)

after annealing was characterized by lifetime tester, as presented in Fig. 2(c). The  $\tau$  of the as-deposited sample is only 3.18  $\mu\text{s}$ , similar to that of a naked Si wafer (1.7  $\mu\text{s}$ ), indicating the rather poor surface passivation effect of the as-deposited  $\text{SiO}_2$  film. However, the surface passivation ability of the LPD  $\text{SiO}_2$  film can be effectively improved by the annealing process. The  $\tau$  of the post-annealed samples gradually increases with the annealing temperature from 500°C to 700°C and then keeps a value of around 35  $\mu\text{s}$  at the annealing temperature of 700°C to 900°C. This improved  $\tau$  may be attributed to the reconstruction of the Si/SiO<sub>2</sub> interface which makes the dangling bonds on Si wafer surface effectively passivated. However, the passivation effect of the  $\text{SiO}_2$  film begins to decrease at a higher annealing temperature of 1000°C. To better understand the function of post-annealing, the silicon wafers were also passivated by other wet oxidation methods ( $\text{H}_2\text{O}_2$  oxidation and  $\text{HNO}_3$  oxidation) and made to go through the same annealing process (800°C, 5 min). However, the  $\tau$ s are only 19  $\mu\text{s}$  and 8  $\mu\text{s}$  for  $\text{HNO}_3$  oxidized wafer and  $\text{H}_2\text{O}_2$  oxidized wafers, respectively. These contrast experiments explicitly demonstrate that there is no thermal oxidation between silicon surface and wet-chemically grown oxides

because the passivation effect is mainly determined by the wet oxidation methods. Overall, a rather high temperature annealing process (700°C–900°C) is of paramount importance for the passivation quality of LPD  $\text{SiO}_2$  film. And the enhanced passivation effect comes from the improved quality of the LPD  $\text{SiO}_2$  film rather than the formation of thermal oxide.

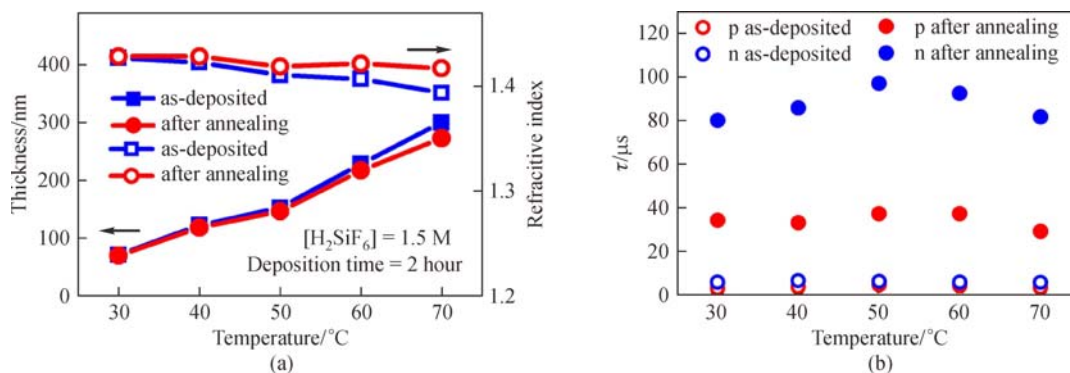
### 3.3 Influence of $\text{H}_2\text{SiF}_6$ concentration

To study the effect of the  $\text{H}_2\text{SiF}_6$  concentration on the properties of the LPD  $\text{SiO}_2$  film, the other growth parameters, e.g., deposition temperature of 50°C, deposition time of 2 h were kept constant. As shown in Fig. 3(a), the thickness of the  $\text{SiO}_2$  film first increases with the  $\text{H}_2\text{SiF}_6$  concentration up to 1.5 M, suggesting an increased deposition rate in the concentration range. The reason for this is that the reversible reaction is promoted to the right side with increased  $\text{H}_2\text{SiF}_6$  concentration, thus facilitating the precipitation of  $\text{SiO}_2$ , as expressed in Eq. (1). Then the thickness of the  $\text{SiO}_2$  film remains around 150 nm in the range of  $\text{H}_2\text{SiF}_6$  concentrations from 1.5 M to 2.0 M, achieving the maximum deposition rate of 75 nm/h.



**Fig. 3** Influence of  $\text{H}_2\text{SiF}_6$  concentration on LPD  $\text{SiO}_2$  film

(a) Dependence of the thickness and the refractive index of the as-deposited LPD  $\text{SiO}_2$  film and the post-annealed LPD  $\text{SiO}_2$  film on  $\text{H}_2\text{SiF}_6$  concentration; (b) influence of  $\text{H}_2\text{SiF}_6$  concentration on  $\tau$  of the as-deposited and the post-annealed n-type and p-type samples



**Fig. 4** Influence of growth temperature on LPD  $\text{SiO}_2$  film

(a) Dependence of the thickness and refractive index of the as-deposited and post-annealed LPD  $\text{SiO}_2$  films on the growth temperature; (b) Influence of growth temperature on  $\tau$  of the as-deposited and post-annealed n-type and p-type samples

However, as further increasing the H<sub>2</sub>SiF<sub>6</sub> concentration to 2.25 M, the film thickness decreases dramatically, which is ascribed to the insufficient H<sub>2</sub>O (as a reactant) at an excessively high H<sub>2</sub>SiF<sub>6</sub> concentration. From the result above, it can be inferred that the concentration of H<sub>2</sub>SiF<sub>6</sub> has a great influence on the deposition rate of the LPD SiO<sub>2</sub> film and a concentration of 1.5 M may be a good choice as the growth condition considering both the process time and material cost. Regarding the refractive index, the value of the LPD SiO<sub>2</sub> film is between 1.41 and 1.43 (except the one grown with 0.5 M) as presented in Fig. 3(a), approaching to that of the thermally grown SiO<sub>2</sub> [21]. After annealing at 800°C, the refractive index of the LPD SiO<sub>2</sub> films is slightly increased while the thickness is decreased. The changes of the film thickness and the refractive index suggest the reconstruction of the LPD SiO<sub>2</sub> film, which may be largely caused by the decreased chemical bond of -OH and -F, incorporate in the as-deposited LPD SiO<sub>2</sub> films.

Figure 3(b) shows the  $\tau$  of the samples with annealing and without annealing as a function of the H<sub>2</sub>SiF<sub>6</sub> concentration. No matter it is n-type textured Si wafers or p-type polished Si wafers, the surface passivation quality of the as-deposited SiO<sub>2</sub> film is rather poor but dramatically enhanced after annealing at 800°C for 5 min in atmosphere, and does not exhibit much dependence on the concentration of H<sub>2</sub>SiF<sub>6</sub>. The excellent passivation effect on both n-type and p-type wafers suggests the chemical passivation mechanism of the LPD SiO<sub>2</sub> film. According to Eq. (2) [22],

$$\frac{1}{\tau} = \frac{1}{\tau_b} + \frac{2S}{W}, \quad (2)$$

where  $\tau_b$  is the bulk lifetime of silicon wafer and  $\tau$  is the measured effective minority carrier lifetime,  $W$  is the thickness of the silicon wafer, the  $\tau_b$ s of n-type and p-type wafers are respectively obtained by iodine-ethanol passivation method and assuming the  $\tau_b$ s are the same for each group of wafers, the average effective surface recombination velocities ( $S$ ) of the LPD SiO<sub>2</sub> passivated n-type and p-type silicon wafers are calculated to be 110 and 260 cm/s, respectively. Obviously, the LPD SiO<sub>2</sub> film has a better passivation effect on the n-type wafer, even though its surface is textured, demonstrating that field-effect passivation also plays an important role in the passivation mechanism of LPD SiO<sub>2</sub> film.

### 3.4 Influence of growth temperature and application in solar cells

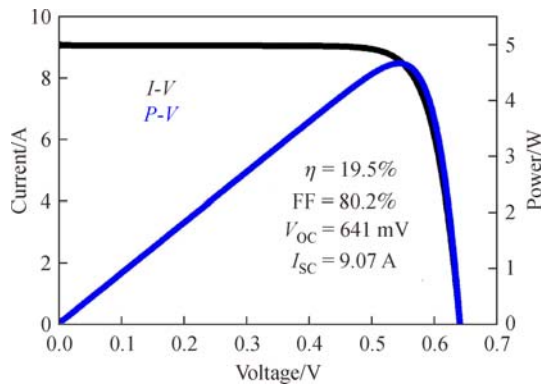
Keeping the H<sub>2</sub>SiF<sub>6</sub> concentration of 1.5 M and the deposition time of 2 h, the influence of growth temperature on the properties of the LPD SiO<sub>2</sub> films was further studied in detail. As shown in Fig. 4(a), the thickness of the LPD SiO<sub>2</sub> films almost increases linearly with increasing

temperature from 30°C to 70°C. Because the hydrolysis of H<sub>2</sub>SiF<sub>6</sub> is an endothermic reaction and the solubility of SiO<sub>2</sub> in H<sub>2</sub>SiF<sub>6</sub> decreases with the increased temperature, a higher growth temperature accelerates the precipitation of SiO<sub>2</sub> [8]. On the contrary, the refractive index of the SiO<sub>2</sub> films decreases with increased growth temperature. After annealing at 800°C for 5 min in atmosphere, all the SiO<sub>2</sub> films grown at different temperatures become thinner and denser. In addition, the influence of growth temperature on the refractive index is weakened. Though the deposition rate increases with growth temperature, the thermal cost also increases. Considering the tradeoff between thermal cost and deposition rate, the growth temperature of 50°C may be the optimal choice. Figure 4(b) illustrates the dependence of the  $\tau$  of the p-type and n-type samples with and without annealing on growth temperature. After annealing, the  $\tau$  of the p-type Si wafer passivated by LPD SiO<sub>2</sub> film reaches around 35  $\mu$ s (except the one with the deposition temperature of 70°C), and that of the n-type samples is between 80  $\mu$ s and 97  $\mu$ s, both suggesting a good passivation quality of the LPD SiO<sub>2</sub> films. Similar to the dependence on the concentration of H<sub>2</sub>SiF<sub>6</sub>, the influence of growth temperature on the surface passivation effect is also not prominent.

All of the above results indicate that the prepared LPD SiO<sub>2</sub> film is suitable to be applied to silicon solar cells as an excellent surface passivation layer. Although the LPD SiO<sub>2</sub> film with a thickness of above 100 nm can be deposited to make it simultaneously act as an antireflection and passivation layer, it is too thick to be punctured by the silver paste, thereby leading to a severe contact problem. Therefore, in this study, the thickness of the SiO<sub>2</sub> film of about 5 nm was controlled to use it only as a passivation layer and SiN<sub>x</sub> layer was still required to be deposited as the antireflection layer. It is worth mentioning that the LPD SiO<sub>2</sub> film with such thickness has almost an equal passivation effect to that with the thickness of several tens of nanometers. Furthermore, the SiN<sub>x</sub> layer also has a contribution to the passivation through field-effect passivation. As a result, as shown in Fig. 5, the conversion efficiency of the LPD SiO<sub>2</sub> passivated solar cells reaches as high as 19.5% with an open circuit voltage of 641 mV, a short circuit current of 9.07 A and a fill factor of 80.2%.

## 4 Conclusions

The high quality SiO<sub>2</sub> film was successfully grown by a novel LPD technique and applied to silicon solar cells as the surface passivation layer. Compared to the common deposition method of SiO<sub>2</sub> film, the LPD is superior with an ultra-low growth temperature (50°C), a relative high deposition rate (75 nm/h), simple equipment and convenient operation. The prepared LPD SiO<sub>2</sub> film exhibits perfect coverage on textured silicon surface with high



**Fig. 5** *I-V* and *P-V* characteristics of the LPD SiO<sub>2</sub> passivated silicon solar cell on a large scale (156 mm × 156 mm)

elemental purity. The passivation quality of the LPD SiO<sub>2</sub> film is strongly improved with increased refractive index after annealing at a temperature higher than 700°C. It is important to find out that the LPD SiO<sub>2</sub> film can effectively passivate both the p-type and n-type Si wafers but results in a lower surface recombination velocity on n-type Si wafers, revealing the simultaneous existence of chemical passivation and field-effect passivation. In addition, although the deposition parameters have little influence on the passivation effect, they affect the deposition rate. An H<sub>2</sub>SiF<sub>6</sub> concentration of 1.5 M and a deposition temperature of 50°C are proposed as the optimal deposition parameters based on the consideration of deposition rate and preparation cost. Finally, the LPD SiO<sub>2</sub> film were successfully applied to silicon solar cells as the front surface passivation layer and realized an efficiency of 19.5% on a large scale (156 mm × 156 mm). It is believed that the cost-effective passivation technique presented in this paper opens a new opportunity for high-efficiency silicon solar cells.

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